**Buccal Patches: A Review**

Farheen Fiza*, Bharadwaj Sudhir, Jat R.C, Arjariya Priyanka, Sharma Garima, Ratnakar Deepit, Arjariya Priyanka, Khan Imran, Tiwari Rahul, Rathore Singh Arvind  
Shri ram college of pharmacy Banmore, Morena (M.P.) India

**ABSTRACT**

Drugs that are administered via the buccal mucosa directly enter the systemic circulation, thereby avoiding hepatic first-pass metabolism. Therefore, this administration route is useful for improving the bioavailability of drugs that are subject to an extensive first-pass effect when delivered orally. For the oral mucosal route of drug administration, various types of dosage forms can be prepared. A sublingual tablet can afford rapid drug absorption and a prompt pharmacological effect; however, the duration of delivery is short owing to the inevitable loss of a large proportion of the administered dose due to swallowing. To avoid such losses, a patch can be formulated that is located on the buccal mucosa of the oral cavity. However, this approach is limited by the thicker dimensions of the buccal membrane compared to the others that line the oral cavity, and constraints impelled by the delivery system itself (the amount of drug reaching the systemic circulation is limited by the area of the mucosa that the patch covers, which, for patient comfort reasons, is relatively small). Direct access to the systemic circulation through the internal jugular vein bypasses drugs from the hepatic first pass metabolism leading to high bioavailability.

**Keywords**

Buccal mucosa, hepatic first-pass metabolism, patch covers.

Please cite this article in press as Fiza Farheen et.al. Buccal Patches: A Review. Indo American Journal of Pharm Research.2013:3(4).
Introduction
Buccal route of drug delivery is a good alternative, amongst the various routes of drug delivery.[1] Buccal drug delivery is most advantageous because it abundant blood supply in buccal mucosa, bypassing the hepatic first-pass effect and accessibility.[2] However, peroral administration of drugs has disadvantages such as hepatic first pass metabolism and enzymatic degradation within the GI tract, that prohibit oral administration of certain classes of drugs especially peptides and proteins. Consequently, other absorptive mucosae are considered as potential sites for drug administration.[3] Oral cavity has been investigated for number of applications including the treatment of periodontal disease bacterial and fungal infection, aphthous and dental stomatitis. Over the last two decades mucocadhesion has become of interest for its systemic delivery by retaining a formulation intimate contact with buccal cavity.[4] The term bio adhesion has been used to define the attachment of a synthetic natural macromolecule to a biological tissue for an extended period of time. When a substrate is a mucosal system adheres and interacts primarily with the mucus layer, this phenomenon being referred to as mucoadhesion.[5] The adhesive properties of such drug delivery platforms can reduce the enzymatic degradation due to the increased intimacy between the delivery vehicle and the absorbing membrane.[6] The use of mucoadhesive polymers in buccal drug delivery has a greater application. Various mucoadhesive devices, including tablets, films, patches, disks, strips, ointments and gels, have recently been developed. However, buccal patch offer greater flexibility and comfort than the other devices. In addition, a patch can circumvent the problem of the relatively short residence time of oral gels on mucosa, since the gels are easily washed away by saliva. Buccal route of drug delivery provides the direct access to the systemic circulation through the jugular vein bypassing the first pass hepatic metabolism leading to high bioavailability.[7]

Advantages
- Bypass of the gastrointestinal tract and hepatic portal system, increasing the bioavailability of orally administered drugs that otherwise undergo hepatic first metabolism.
- Improved patient compliance due to the elimination of associated pain with injections; administration of drugs in unconscious or incapacitated patients.
- Sustained drug delivery.
- A relatively rapid onset of action can be achieved relative to the oral route, and the formulation can be removed if therapy is required to be discontinued.
- Increased ease of drug administration.[8]

Disadvantages
- Limited absorption area- the total surface area of the membranes of the oral cavity available for drug absorption is 170 cm2 of which ~50 cm2 represents non-keratinized tissues, including buccal membrane.[9]
- The barriers such as saliva, mucus, membrane coating granules, basement membrane etc retard the rate and extent of drug absorption through the buccal mucosa.[10]
- Continuous secretion of the saliva(0.5-2 l/day)leads to subsequent dilution of the drug.[11]
- The hazard of choking by involuntarily swallowing the delivery system is a concern.
- Swallowing of saliva can also potentially lead to the loss of dissolved or suspended drug and ultimately the involuntary removal of the dosage form.[12]
Structure & Design of Buccal Dosage Form

Structure and design: Drug delivery designed for the buccal mucosa contains a polymeric adhesive component. When in contact with the saliva, the adhesive attaches to the mucosa causing immediate and rapid drug delivery. Transmucosal drug delivery systems can be unidirectional or bi-directional. Unidirectional patches release the drug only into the mucosa, while bi-directional patches release the drug in both the mucosa and the mouth. The buccal patch is designed in either a matrix configuration with drug, adhesive, and additives mixed together, or a reservoir system that contains a cavity for the drug and additives separate from the additives. An impermeable backing is applied to control the direction of drug delivery, to reduce patch deformation and disintegration while in the mouth; and to prevent drug loss. Additionally, the patch can be constructed to undergo minimal degradation in the mouth, or can be designed to dissolve almost immediately.

Buccal dosage form for buccal delivery: In the past decades, till now, different drug delivery systems intended for buccal administration have been developed. The most common buccal dosage forms are tablets and patches. Such type of form must be of a small size and a suitable geometry so as to not interfere with physiological function of the mouth, even after their hydration in the oral cavity. One of the requirements is that they do not adhere too tightly because it is undesirable to exert too much force to remove the formulation/dosage form after use, otherwise the mucosa could be injured. An alternative is the use of formulations that dissolve or disintegrate completely during the application period. Moreover, in the case of Transmucosal administration, Drug release should be unidirectional (towards the mucosa), and the release into the saliva should be avoided.

Matrix type: The buccal patch designed in a matrix configuration contains drug, adhesive, and additives mixed together.

Reservoir types: The buccal patch designed in a reservoir system contains a cavity for the drug and additives separate from the adhesive. An impermeable backing is applied to control the direction of drug delivery; to reduce patch deformation and disintegration while in the mouth; and to prevent drug loss. Additionally, the patch can be constructed to undergo minimal degradation in the mouth, or can be designed to dissolve almost immediately.

Patches: Patches are laminated and generally consist of an impermeable backing layer and a drug-containing layer that has mucoadhesive properties and from which the drug is released in a controlled manner. Moreover, buccal patches for systemic delivery of tyrotropin-releasing hormone, octreotide, oxytocin, buserelin, calcitonin and leuenkephalin have been studied.

Novel drug delivery system: Novel drug delivery systems, such as lipophilic gel, buccal spray and phospholipids vesicles have been recently proposed to deliver peptides via the buccal route. A novel liquid aerosol formulation (Oralin, Generex Biotechnology) has already been developed. This system allows precise insulin dose delivery via a metered dose inhaler in the form of fine aerosolized droplets directed into the mouth. This oral aerosol formulation is rapidly absorbed through the buccal mucosal epithelium, and it provides the plasma insulin levels necessary to control postprandial glucose rise in diabetic patients. This novel, pain-free, oral insulin formulation has a number of advantages including rapid absorption, a simple (user-friendly) administration technique, precise dosing control (comparable to injection within one unit) and bolus delivery of drug.[13]
Figure 1. Schematic representation of different matrix tablets for buccal delivery. Arrows indicate the direction of drug release.

**Buccal drug delivery system**
A delivery system designed to deliver drug systemically or locally via buccal mucosa. Buccal delivery refers to the drug release which can occur when a dosage form is placed in the outer vestibule between the buccal mucosa and gingival.[14]

**Buccal dosage forms**
- **Buccal mucoadhesive tablets:**
  Buccal mucoadhesive tablets are dry dosage form that have to be moistened prior to placing in contact with buccal mucosa. Example: a double layer tablet, consisting of adhesive matrix layer of hydroxyl propyl cellulose and polyacrylic acid with an inner core of cocoa butter containing insulin and a penetration enhancer (sodium glycocholate).

- **Patches and Films:**
  Buccal patches consist of two laminates, with an aqueous solution of the adhesive polymer being cast onto an impermeable backing sheet, which is then cut into the required oval shape. A novel mucosal adhesive film called “Zilactin” – consisting of an alcoholic solution of hydroxy propyl cellulose and
three organic acids. The film which is applied to the oral mucosal can be retained in place for at least 12 hours even when it is challenged with fluids.[15]

- **Semisolid Preparations (Ointment and Gels):**
  Bioadhesive gels or ointment have less patient acceptability than solid bioadhesive dosage form, and most of the dosage forms are used only for localized drug therapy within the oral cavity. One of the original oral mucoadhesive delivery systems—“orabase”- consists of finely ground pectin, gelatin and sodium carboxy methylcellulose dispersed in a poly (ethylene) and a ground pectin, gelatin and sodium carboxy methylcellulose dispersed in poly (ethylene) and a mineral oil gel base, which can be maintained at its site of application for 15-150 minutes.[16]

- **Powders:**
  Hydroxypropyl cellulose and beclomethasone in powder form when sprayed onto the oral mucosa of rats, a significant increase in the residence time relative to an oral solution is seen, and 2.5% of beclomethasone is retained on buccal mucosa for over 4 hours.[17]

**Mechanism of buccal absorption**
Buccal drug absorption occurs by passive diffusion of the nonionized species, a process governed primarily by a concentration gradient, through the intercellular spaces of the epithelium. The passive transport of non-ionic species across the lipid membrane of the buccal cavity is the primary transport mechanism. The buccal mucosa has been said to be a lipoidal barrier to the passage of drugs, as is the case with many other mucosal membrane and the more lipophilic the drug molecule, the more readily it is absorbed. The dynamics of buccal absorption of drugs could be adequately described by first order rate process. Several potential barriers to buccal drug absorption have been identified. Dearden and Tomlison (1971) pointed out that salivary secretion alters the buccal absorption kinetics from drug solution by changing the concentration of drug in the mouth. The linear relationship between salivary secretion and time is given as follows:

\[
\frac{\text{dm}}{\text{dt}} = \frac{KC}{V \cdot V_t}
\]

where,
- \( M \) – Mass of drug in mouth at time \( t \)
- \( K \) – Proportionality constant
- \( C \) – Concentration of drug in mouth at time
- \( V \cdot V_t \) – The volume of solution put into mouth cavity and \( V_t \) Salivary secretion rate

**Factors affecting buccal absorption**
The oral cavity is a complex environment for drug delivery as there are many interdependent and independent factors which reduce the absorbable concentration at the site of absorption.

1. **Membrane Factors**
This involves degree of keratinization, surface area available for absorption, mucus layer of salivary pellicle, intercellular lipids of epithelium, basement membrane and lamina propria. In addition, the absorptive membrane thickness, blood supply/lymph drainage, cell renewal and enzyme content will all contribute to reducing the rate and amount of drug entering the systemic circulation.

2. **Environmental Factors**
A.) **Saliva:** The thin film of saliva coats throughout the lining of buccal mucosa and is called salivary pellicle or film. The thickness of salivary film is 0.07 to 0.10 mm. The thickness, composition and movement of this film
affect the rate of buccal absorption.

B.) Salivary glands: The minor salivary glands are located in epithelial or deep epithelial region of buccal mucosa. They constantly secrete mucus on surface of buccal mucosa. Although, mucus helps to retain mucoadhesive dosage forms, it is potential barrier to drug penetration.

C.) Movement of buccal tissues: Buccal region of oral cavity shows less active movements. The mucoadhesive polymers are to be incorporated to keep dosage form at buccal region for long periods to withstand tissue movements during talking and if possible during eating food or swallowing.[18]

Composition of buccal patches

A. Active Pharmaceutical ingredient (API):
The buccal film technology has the potential for delivery of variety of APIs. However since the size of the dosage form has limitation, high dose molecules are difficult to be incorporated in buccal film. Generally 5%-w/w to 30%-w/w of active pharmaceutical ingredients can be incorporated in the buccal patches.[19]

B. Polymers (adhesive layer):
Polymer hydration and swelling properties probably play the main role. The polymer hydration and consequently the mucus dehydration could cause an increase in mucous cohesive properties that promote mucoadhesion. Swelling should favor polymer chain flexibility and interpenetration between polymer and mucin chains. So, depending on the type of formulation, polymers with different characteristics have to be considered.
Examples: Hydroxy ethylcellulose, hydroxypropyl cellulose, polyvinyl pyrrolidone, polyvinyl alcohol, carbopol and other mucoadhesive polymers.[20]

C. Diluents:
Lactose DC is selected as diluent for its high aqueous solubility, its flavouring characteristics, and its physico-mechanical properties, which make it suitable for direct compression. other example : microcrystalline starch and starch.

D. Sweetening agents:
Sucralose, aspartame, mannitol, etc.

E. Flavouring agents:
Menthol, vanillin, clove oil, Peppermint oil, cinnamon oil, spearmint oil, oil of nutmeg are examples of flavor oils while vanilla, cocoa, coffee, chocolate and etc.[21]

F. Backing layer:
Ethyl cellulose, etc.

G. Penetration enhancer:
Cyano acrylate, EDTA, Citric acid etc.

H. Plasticizers:
PEG-100, 400, propylene glycol, etc.[22]

Method of preparation
Two methods used to prepare adhesive patches include,

Solvent casting:
In this, all patch excipients including the drug codispersed in an organic solvent and coated Onto a sheet of release liner. After solvent evaporation, a thin layer of the protective backing material is laminated onto the sheet of coated release liner to form a laminate that is die-cut to form patches of the desired size and geometry.
The solvent casting method is simple, but suffers from some disadvantages, including long processing time, high cost, and environmental concerns due to the solvents used. These drawbacks can be overcome by the hot-melt extrusion method.[23]

**Direct milling:**
In this, patches are manufactured without the use of solvents (solvent-free). Drug and excipients are mechanically mixed by direct milling or by kneading, usually without the presence of any liquids. After the mixing process, the resultant material is rolled on a release liner until the desired thickness is achieved. An impermeable backing membrane may also be applied to control the direction of drug release, prevent drug loss, and minimize deformation and disintegration of the device during application period.[24]

While there are only minor or even no differences in patch performance between patches fabricated with the two processes, the solvent-free process is preferred because there is no possibility of residual solvents and no associated solvent-related health issues.[25]
List of drug delivered via buccal route

In an effort to determine the feasibility of buccal route as a novel route of drug delivery, several drugs have been studied. The variation in class of compounds illustrates that the pharmaceutical industries have an alternative and novel routes of administration for existing drugs.[26]

Active Ingredients:
- Acitretin
- Acyclovir
- Arecoline
- Buprenorpine
- Carbamazepine
- Chitosan
- Chlorpheniramine maleate
- Metronidazole
- Morphine sulphate
- Nicotine
- Nifedipine
- Omeprazole
- Oxytocin
- Piroxicam
- Ergotamine tartrate (etc).

Evaluation

Surface pH:
The surface pH of the buccal patch was determined in order to investigate the possibility of any side effects in vivo. As an acidic or alkaline pH may cause irritation to the buccal mucosa, it was determined to keep the surface pH as close to neutral as possible. A combined glass electrode was used for this purpose. The patches were allowed to swell by keeping it in contact with 1 ml of distilled water (pH 6.5 ± 0.05) for 2 hours at room temperature, and pH was note down by bringing the electrode in contact with the surface of the patch and allowing it to equilibrate for 1 minute.[27]

Swelling studies:
Weight and area increase due to swelling were measured. Weight increase due to swelling: A drug-loaded patch of 1x1 cm² was weighed on a preweighed cover slip. It was kept in a petridish and 50 ml of phosphate buffer, pH 6.6 was added. After every five minutes, the cover slip was removed and weighed up to 30 minutes. The difference in the weights gives the weight increase due to absorption of water and swelling of patch. Area increase due to swelling: A drug loaded patch size of 1x1 cm² was cut and placed in a petridish. A graph paper was placed beneath the petridish, to measure the increase in the area. 50ml of phosphate buffer, pH 6.6, was poured into the petridish. An increase in the length and breadth of the patch was noted at 5 min intervals for 60 min and area was calculated.[28]

The percent swelling, %S, was calculated using the following equation:

\[ \frac{X_t - X_o}{X_o} \times 100 \]
Where $X_t$ is the weight or area of the swollen patch after time $t$
$X_0$ is the original patch weight or area at zero time

**Thickness measurements:**
The thickness of each film is measured at five different locations (centre and four corners) using an electronic digital micrometer.

**Thermal analysis study:**
Thermal analysis study is performed using differential scanning calorimeter (DSC).

**Morphological characters:**
Morphological characters are studied by using scanning electron microscope (SEM).[29]

**Palatability test:**
Palatability study is conducted on the basis of taste, after bitterness and physical appearance. All the batches are rated A, B and C grades as per the criteria. When the formulation scores at least one A grade, formulation is considered as average. When the formulation score two a grade then it would be considered as good and the one with all three A grade it would be the very good formulation.[30]

Grades: A = very good, B = good, C = poor

**Folding endurance:**
The test is performed by repeated folding of the film at the same place until film failure. A maximum of 300 times is sometimes reported as a limit to the test, and the value is reported as the number of times the film can be folded prior to rupture.[31]

**In vitro drug release:**
The United States Pharmacopeia (USP) XXIII rotating paddle method used to study the drug Release from the bilayered and multilayered patches. The dissolution medium consisted of phosphate buffer pH 6.8. the release was performed at $37 \degree C \pm 0.50 \degree C$, with a rotation speed of 50 rpm. The backing layer of buccal patches attached to the glass disk with instant adhesive (cyanoacrylate adhesive). The disk was allocated to the bottom of the dissolution vessel. Samples (5ml) were withdrawn at predetermined time intervals and replaced with fresh medium. The samples filtered through whatman filter paper and analyzed after appropriate dilution by UV spectrophotometry at suitable nm.[32]

**In vitro drug permeation:**
The in vitro buccal drug permeation study of Drugs through the buccal mucosa (sheep and rabbit) performed using Keshary-Chien/Franz type glass diffusion cell at $37\degree C \pm 0.2\degree C$. Fresh buccal mucosa mounted between the donor and receptor compartments. The buccal tablet was placed with the core facing the mucosa and the compartments clamped together. The donor compartment filled with 1 ml of phosphate buffer pH 6.8. The receptor compartment was filled with phosphate buffer pH 7.4, and the hydrodynamics in the receptor compartment maintained by stirring with a magnetic bead at 50 rpm. A one ml sample can be withdrawn at predetermined time intervals and analyzed for drug content at suitable nm using a UV spectrophotometer.[33]

**Stability study in Human saliva:**
Stability study of fast dissolving films is carried out for all the batches according to ICH guidelines. After predetermined time intervals, the films are evaluated for the drug content, disintegration time and physical appearance. The stability study of optimized mucoadhesive patch formulation was performed at 400C, 37 ±50C & 75±5% RH for three months. The value of all parameter after three months remain same as their values and
minor changes occur in value of volume entrapment efficiency, % elongation & % drug release after 8 hour which was considerable.[34-35]

**Ex vivo mucoadhesive strength:**
A modified balance method used for determining the ex vivo mucoadhesive strength. Fresh buccal mucosa (sheep and rabbit) obtained, used within 2 hours of slaughter. The mucosal membrane separated by removing underlying fat and loose tissues. The membrane washed with distilled water and then with phosphate buffer pH 6.8 at 37°C. The buccal mucosa cut into pieces and washed with phosphate buffer pH 6.8. A piece of buccal mucosa was tied to the glass vial, which was filled with phosphate buffer. The two sides of the balance made equal before the study, by keeping a 5 g weight on the right-hand pan. A weight of 5 g was removed from the right-hand pan, which lowered the pan along with the tablet over the mucosa. The balance was kept in this position for 5 minutes contact time. The water (equivalent to weight) was added slowly with an infusion set (100 drops/min) to the righthand pan until the tablet detached from the mucosal surface. This detachment force gave the mucoadhesive strength of the buccal tablet in grams. The glass vial was tightly fitted into a glass beaker (filled with phosphate buffer pH 6.8, at 37°C ±1°C) so that it just touched the mucosal surface. The buccal tablet was stuck to the lower side of a rubber stopper with cyanoacrylate adhesive.[36]

**Conclusion:**
Buccal region provides a convenient route of administration for both local and systemic drug actions. Controlled buccal drug delivery systems, where the drug delivery is directed towards buccal mucosa by protecting the local environment is also gaining interest. Currently solid dosage forms, liquids and gels applied to oral cavity are commercially successful. The future direction of buccal adhesive drug delivery lies in vaccine formulations and delivery of small proteins/peptides. Microparticulate bioadhesive systems are particularly interesting as they offer protection to therapeutic entities as well as the enhanced absorption that result from increased contact time provided by the bioadhesive component. Exciting challenges remain to influence the bioavailability of drugs across the buccal mucosa. Many issues are yet to be resolved before the safe and effective delivery through buccal mucosa. In mucoadhesive placebo buccal patches we can use any potent drugs which fulfill the criteria for buccal patch as drug delivery system.

**References**
7. Shidhaye SS, Mucoadhesive bilayered patches for administration of sumatriptan, AAPS pharm sci tech, 2009, 9(3).
27. Peppas NA, Buri PA. Surface, interfacial and molecular aspects of polymer bioadhesion


