Structure Types of Kidney Stones and Their Susceptibility to Shock Wave Fragmentation

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1. INTRODUCTION
Urolithiasis takes a leading place in the structure of urological diseases. Its prevalence in the modern population, according to various epidemiological studies, is 5-13%, depending on the region (1-3). The modern approach in the treatment of urolithiasis involves the use of non-invasive and minimally invasive techniques based on the stone fragmentation, among which shock wave lithotripsy (SWL) is considered as the first-line treatment for kidney stones < 2 cm and proximal ureter stones (4-6).

One of the factors influencing the effectiveness of SWL is the mineral composition of the stone (7, 8).

However, there are studies reporting that stones with the same mineral composition have different character of fragmentation when exposed to shock waves (9, 10). The possible cause of this phenomenon may be the special features of stones determined by a variable structural state of their mineral components.

2. AIM
In this work, we have studied the microstructure and mineral composition of kidney stones, as well as their predisposition to fragmentation by shock waves.
2. MATERIALS AND METHODS

We have studied the microstructure and mineral composition of kidney stone fragments obtained from 87 patients (27 females, 60 males) aged 26-63 years (average age 41.08±8.62) with solitary renal stones after SWL done with Dornier Compact Sigma Lithotripter (Dornier Medtech, Germany) at V. I. Shapoval Regional Medical Clinical Center of Urology and Nephrology, Kharkiv, Ukraine. The stone microstructure was assessed by the method of crystal optical analysis in conjunction with immersion liquids using the polarizing microscope Polam 211L, LOMO (Russia).

With the use of the eyepiece graticule, based on the principles of quantitative analysis of microscopic images (11), the quantitative parameters of the amorphous and crystalline phases of the stone composition such as volume fraction of the amorphous phase (VFAP) and volume fraction of the crystalline phase (VFCP) were calculated. The linear dimensions, shape, color, and transparency level of the crystalline elements were also assessed.

The stone mineral composition was measured by infrared (IR) spectroscopy method using the IR spectrometer IRS-29 (LOMO, Russia) with the spectral range of 4000-400 cm⁻¹. Powdered samples obtained by grinding the fragments of urinary tract stones in agate mortars to a particle size of ~ 1-10 μm were studied. The samples were prepared from a mixture of potassium bromide as a matrix (99%) and the test substance (1%). A 100-mg aliquot of the resulting homogenized powder was then pressed into a transparent pellet. To exclude the matrix absorption bands, a pure potassium bromide pellet, preliminarily dried at 180°C during 10 hours, was placed in the sample compartment of the device.

The calibration was performed according to the spectrum of polystyrene with known frequencies of absorption maxima. The adjustment averaged 5 to 10 cm⁻¹. The mineral composition of the stone was evaluated based on the identification of the infrared absorption bands which were specific to certain chemical compounds (12, 13). The intensity of the absorption bands, the characteristics of the maxima, and the transmission level were also determined.

At the final stage of the study, the initial parameters and susceptibility to shock wave fragmentation of stones with different structural features were assessed based on the retrospective analysis of the patients’ medical records. The initial stone parameters were determined using non-contrast computed tomography (NCCT) with Toshiba Aquilion 16 CT scanner (Japan), performed in all patients before SWL. The maximum stone length (MSL) and mean stone density (MSD) were assessed. The maximum linear size of the stone in the axial or coronal plane was considered as the MSL (14). The MSD was calculated as the mean density index in Hounsfield units, measured in the plane where the elliptical region of interest included the largest cross-sectional area of the stone, excluding adjacent soft tissue (15). The stone susceptibility to shock wave fragmentation was assessed by the total number of shock waves (SWS) during all lithotripsy sessions required for complete stone fragmentation and achieving the “stone free” status meaning absolute clearance or residual stone fragments less than 4 mm according to the NCCT confirmation at the end of treatment.

Statistical data processing was performed using Microsoft Excel 2016 spreadsheets and Statistica 10 (StatSoft, USA). An intergroup comparison of three independent samples was performed using the Kruskal-Wallis test. The quantitative variables, that showed statistically significant differences (p<0.05) according to the Kruskal-Wallis criterion, were additionally analyzed using a post hoc Mann-Whitney U test.

The direction and strength of the relationship between the variables were evaluated using Spearman’s correlation coefficient.

3. RESULTS

The crystal optical analysis of the urinary stone samples revealed the presence of amorphous and crystalline phases, either isolated or combined with each other.

The amorphous phase, when observed in an immersion preparation, presented as a translucent or opaque mass, composed by grayish-brown clusters of an irreg-
ular shape with indistinct edges, of 30-120 μm in size (Figure 1).

The crystalline phase was characterized by various structural elements corresponding to different evolutionary stages of crystal formation: globules of 10–50 μm in size with dark edges and transparent crystalline substance in the center (crystallization nuclei) (Figure 2); transparent, translucent and opaque grains of 20-80 μm with various color intensity, from light-beige to black (Figure 3).

According to the quantitative ratio between the amorphous and crystalline phases of the stone structure, as well as the available data on the evolutionary principles of biominerals formation involving the consecutive crystallization of the amorphous phase (16-18), three structure types of kidney stones were distinguished.

Structure type A: an amorphous-crystalline structure with the predominant amorphous phase (VFAP > 50 vol%). The crystalline phase was formed of crystalline nuclei and fine crystalline mass (up to 30 μm) without clear differentiation (Figure 4).

Structure type B: an amorphous-crystalline structure with the predominant crystalline phase (VFCP > 50 vol%). The fine crystalline mass was formed of crystal grains of varying transparency (Figure 5).

Structure type C: a crystalline structure without the amorphous phase. The fully crystalline phase (VFCP = 100 vol%) was formed of crystals in the shape of grains with the structural characteristics depending on the mineral type (Figure 6).

Out of all 87 samples analyzed, structure type A, B and C stones were identified in 22 (25.29%), 39 (44.83%) and 26 (29.88%) patients, respectively.

Infrared spectroscopy of various structure type stones revealed a wide range of chemical compounds. The most common chemical compounds were calcium oxalate (CaOx) in the form of whewellite (calcium oxalate mono-

<table>
<thead>
<tr>
<th>Number of mineral components</th>
<th>Qualitative mineral composition of the stone</th>
<th>Structure type A n=22</th>
<th>Structure type B n=39</th>
<th>Structure type C n=26</th>
</tr>
</thead>
<tbody>
<tr>
<td>1 component, n=21</td>
<td>CaOx</td>
<td>3 (13.64%)</td>
<td>4 (10.26%)</td>
<td>4 (15.38%)</td>
</tr>
<tr>
<td></td>
<td>Uric acid</td>
<td>2 (9.09%)</td>
<td>4 (10.26%)</td>
<td>3 (11.54%)</td>
</tr>
<tr>
<td></td>
<td>Ammonium urate</td>
<td>0 (2.56%)</td>
<td>1 (16.26%)</td>
<td>0 (0.00%)</td>
</tr>
<tr>
<td>2 components, n=52</td>
<td>CaOx + CaP</td>
<td>12 (54.54%)</td>
<td>20 (51.28%)</td>
<td>11 (42.31%)</td>
</tr>
<tr>
<td></td>
<td>CaOx + uric acid</td>
<td>2 (9.09%)</td>
<td>4 (10.26%)</td>
<td>3 (11.54%)</td>
</tr>
<tr>
<td>3 components, n=14</td>
<td>CaOx + Ca carbonate + uric acid</td>
<td>0 (2.56%)</td>
<td>1 (16.26%)</td>
<td>0 (0.00%)</td>
</tr>
<tr>
<td></td>
<td>CaOx + CaP + uric acid</td>
<td>3 (13.64%)</td>
<td>5 (12.82%)</td>
<td>4 (15.38%)</td>
</tr>
<tr>
<td></td>
<td>Uric acid + CaP + aragonite</td>
<td>0 (2.56%)</td>
<td>1 (16.26%)</td>
<td>0 (0.00%)</td>
</tr>
</tbody>
</table>

Table 1. Quantitative and qualitative characteristics of the stone structure types
hydrate) or wedellite (calcium oxalate dihydrate), calcium phosphates (CaP) in the form of apatite, hydroxylapatite, fluorapatite, and uric acid. Aragonite, calcium carbonate, and ammonium urate were identified in rare cases. Sixty-six (75.86%) stones had a mixed mineral composition with two or more components (Table 1).

The appearance of the IR spectrum was dependent on the structural state of the minerals composing the stone. The amorphous phase was characterized by absorption bands of low and medium intensity, often with wide or blurred maxima and low transmittance (T=30-50%). The spectral properties of the crystalline phase were the increased number of characteristic absorption bands of medium and high intensity with narrow maxima, as well as high transmittance (T=70-80%) (Figure 7).

Despite the qualitative differences between the IR spectra of the amorphous and crystalline phases, it was not possible to measure their quantity in the stone sample and to determine the stone structure type by the spectral curve pattern.

The features of the kidney stone structure types were evaluated based on the tomographic parameters (MSL, MSD), as well as SWs number characterizing the stone susceptibility to shock wave fragmentation. Comparative analysis of various structure types of kidney stones revealed statistically significant differences in SWs between the study groups. The maximum values of this parameter was observed in the group of patients with structure type C stones. No significant difference between MSL and MSD parameters was seen in the stone samples of different structure types (p>0.05) (Table 2).

Table 2. Comparative analysis of kidney stone structure types Me–median; MSD – mean stone density; MSL – maximum stone length; HU – Hounsfield Units; [Q1-Q3] – interquartile range; SWs – total number of shock waves required for complete stone fragmentation. *–Kruskal-Wallis test; †–significant difference (p<0.01) compared to type A (post hoc analysis using Mann-Whitney U test); ‡–significant difference (p<0.01) compared to type B (post hoc analysis using Mann-Whitney U test).

<table>
<thead>
<tr>
<th>Parameters</th>
<th>Type A</th>
<th>Type B</th>
<th>Type C</th>
<th>P-value*</th>
</tr>
</thead>
<tbody>
<tr>
<td>Me (11-03)</td>
<td>1274.50</td>
<td>1259.00</td>
<td>1466.50</td>
<td>0.13</td>
</tr>
<tr>
<td>MSD (HU)</td>
<td>1274.50</td>
<td>1259.00</td>
<td>1466.50</td>
<td>0.13</td>
</tr>
<tr>
<td>MSL (mm)</td>
<td>11.00</td>
<td>12.00</td>
<td>12.00</td>
<td>0.094</td>
</tr>
<tr>
<td>SWs</td>
<td>2990.00</td>
<td>4200.00</td>
<td>5025.00</td>
<td>&lt;0.01</td>
</tr>
</tbody>
</table>

Figure 6. An immersion photomicrograph of a kidney stone specimen under transmitted light. Structure type C: a crystalline structure represented by large (100-250 μm) transparent and translucent grains of gray apatite (1), among which grains of whewellite with characteristic streaked texture (2) and a fine crystalline mass of calcium phosphate can be encountered (3). VFCP = 100 vol%.

Figure 7. IR spectra of amorphous, amorphous-crystalline and crystalline whewellite in the range of 600-1000 cm⁻¹. Curve 1–crystalline whewellite; Curve 2–amorphous-crystalline whewellite (VFAP~ 30 vol%); Curve 3–amorphous whewellite.

4. DISCUSSION

The current trend in the management of urolithiasis is the use of various types of lithotripsy, among which SWL remains relevant and continues to be considered as the first-line treatment in most patients with kidney and ureter stones (19-21). Among the factors influencing the outcomes of SWL, the composition- and structure-dependent fragility of a calculus appears to be of great importance (7, 22, 23). Therefore, the adequate analysis of a kidney stone should involve a comprehensive assessment of its mineral composition and internal structure. Currently, the IR spectroscopy and X-ray diffraction are widely used as the methods for evaluating the mineral composition of a stone (24-26). However, these methods do not allow us to make a comprehensive assessment of
the calculus structural features.

It is reported that the synchrotron radiation microtomography (SR-μCT) is used for evaluation of the microstructure and mineral composition of kidney stones. However, this method cannot be easily applied to the routine practice, and is only appropriate in the study of rare, non-typical samples (27).

In our work, the stone microstructure was assessed by crystal optical analysis using a polarizing microscope with immersion liquids. Polarization microscopy as a method for analyzing kidney stones has been used by a number of authors to evaluate the stone mineral composition (28, 29). However, the accuracy of component identification in the mixed combinations, especially those containing uric acid, calcium phosphate and purine derivatives, is inferior to spectroscopic methods (29), which is why it has not been widely used in the urolithiasis diagnosis. The use of polarization microscopy is more reasonable in cases of assessment of the volume fraction of the crystalline and amorphous phases. The advantages of this method are its economic efficiency, speed of execution, and the ability to use a small amount of substance (30).

According to modern concepts, the crystallization process passes through the amorphous phase (16-18). In 70.1% of our patients, the amorphous phase in the structure of the urinary stone was present along with the crystalline phase, which was a sign of an incomplete crystallization process, suggesting a relatively young age of a stone. The ratio of the crystalline and amorphous phases of the calculus structure is both characteristic of the temporal parameters of stone formation, and also is a factor determining the degree of stone predisposition to fragmentation by shock waves.

The results of our study show that the stone structure types with a higher level of the VFAP and immature crystalline forms are better disintegrated by SWL, allowing us to consider the stone structure type as the factor impacting the effectiveness of SWL. Further research in the field of urolithiasis diagnosis is apparently needed to find the visual criteria for distinguishing the structure types of urinary stones, which will be the basis for a differentiated approach to choosing a treatment option in different patients.

Based on the data of the crystal optical analysis, the renal stones, according to their structural state depending on the volume ratio between the amorphous and crystalline phases can be divided into three types: structure type A, which are the amorphous-crystalline structure stones with the predominant content of the amorphous phase (> 50 vol%); structure type B, which are the amorphous-crystalline structure stones with the predominant content of the crystalline phase (> 50 vol%); structure type C, which are the stones with fully crystalline structure. The presence of the amorphous phase, as well as immature crystalline forms, indicates an incomplete crystallization process and makes the stone more susceptible to shock wave exposition. The increased VFCP of the stone structure reduces its susceptibility to shock wave fragmentation. Determining the structure type of a stone by crystal optical analysis is a reasonable component of a complex assessment of kidney stones.

REFERENCES


